Discovery of Benzodiazepine Sulfonamide-Based Bombesin Receptor Subtype 3 Agonists and Their Unusual Chirality

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Supporting Information

ABSTRACT: We report herein the discovery of benzodiazepine sulfonamide-based bombesin receptor subtype 3 (BRS-3) agonists and their unusual chirality. Starting from a high-throughput screening lead, we prepared a series of BRS-3 agonists with improved potency and pharmacokinetic properties, of which compound **8a** caused mechanism-based, dose-dependent food intake reduction and body weight loss after oral dosing in diet-induced obese mice. This effort also led to the discovery of a novel family of chiral molecules originated from the conformationally constrained seven-membered diazepine ring.

ing. H N⁻ CF₃ H N⁻

KEYWORDS: Bombesin receptor subtype 3, BRS-3, agonist, obesity, benzodiazepine sulfonamide, planar chirality, atropisomerism

besity is a serious and chronic medical condition that has become a major global health issue. In the United States, obesity rates for adults have doubled, and rates for children have tripled over the last 30 years. Currently, around one-third of U.S. adults and 16% of U.S. children are considered obese.¹ In many cases, excessive body weight is the root cause of subsequent comorbidities, including type 2 diabetes, hypertension, cardiovascular disease, cancer, and arthritis.² The preferred approach to manage obesity calls for lifestyle changes including control of calorie intake and increase in physical activities. However, this approach alone is often insufficient or unsustainable. In the United States, orlistat³ is the only drug approved for the treatment of obesity and has suboptimal tolerability and limited efficacy (the other obesity drug, sibutramine, was recently withdrawn from the market by Abbott Laboratories⁴). Therefore, new antiobesity agents are highly desirable. Our laboratories are interested in pursuing new mechanisms for the treatment of obesity. The involvement of the bombesin receptor subtype-3 (BRS-3) in regulating energy homeostasis has been demonstrated in animal models. Mice lacking functional BRS-3 develop metabolic defects and obesity.^{5,6} On the other hand, a BRS-3 agonist caused food intake reduction and an increase in metabolic rate in established diet-induced obese (eDIO) mice, suggesting that the BRS-3 receptor may serve as a potential target for treating obesity.⁷ Recently, medicinal chemistry efforts to identify small molecule BRS-3 agonists have been reported by others^{8,9} and our laboratories.^{10–14} In this letter, we describe our continued efforts in this area that led to the discovery of a novel class of BRS-3 agonists.

Our efforts started with a high-throughput screening lead (1) as shown in Figure 1. This lead exhibited moderate BRS-3 potency in human and mouse and poor pharmacokinetic properties in the rat ($Cl_p = 164 \text{ mL/min/kg}$, F = 1%, $t_{1/2} = 1.6 \text{ h}$). Our initial aim was to determine whether the diazepine core was optimal. A systematic core structure modification was undertaken, which included moving the pyridine N, breaking the diazepine ring, and saturating the left-hand side phenyl ring. Each case resulted in complete loss of potency. Therefore, the diazepine core was preserved for further structure—activity relationship (SAR) studies that focused on varying the substituents of the aromatic rings.

The synthesis of the diazepine sulfonamide analogues was straightforward (Scheme 1). Condensation of substituted phenyldiamine 2 and 2-chloronicotinic acid 3 under microwave or thermal conditions provided a regioisomeric mixture of 4 and 5, favoring the desired isomer 4 in a ratio of 3:1 to 9:1. This mixture was reduced by borane to diazepine 6, which upon reacting with the sulfonyl chloride afforded diazepine sulfonamide 8. Typically, the undesired regioisomer was carried on to the final step where 7 was separated from 8 by silica gel chromatography.

However, both silica gel chromatography and reverse-phase highperformance liquid chromatography (HPLC) failed to provide an acceptable separation of 7a from 8. Then, we subjected the mixture to a ChiralCel OD column, expecting two peaks corresponding to

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Figure 1. Structures of orlistat, sibutramine, and HTS lead 1.





^{*a*} Reagents and conditions: (a) 2-Butoxyethanol, microwave 180–220 °C, 15 min to 1 h; or thermoheating, 150 °C, 4–6 h. (b) BH₃·THF. (c) CH₂Cl₂, pyr.



Figure 2. Discovery of the atropisomerism.

the two regioisomers. To our surprise, three peaks were present in the HPLC trace. While one peak corresponded to the undesired



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Figure 3. X-ray analysis revealed planar chirality.¹⁷



Figure 4. "SAR" of the atropisomerism. $R_A = H$ (8k) or F (8l), no atropisomerism; 9 did not exhibit atropisomerism. If $R_A > F$, 8 exhibited atropisomerism, e.g., 8i ($R_A = Cl$), 8m ($R_A = OH$), and 10.

regioisomer 7a, the two compounds obtained corresponding to the other two peaks displayed exactly the same ¹H NMR but opposite rotation (8a, $[\alpha]_{\rm D} = -157^{\circ}$; 8b, $[\alpha]_{\rm D} = +155^{\circ}$), suggesting an enantiomeric pair (Figure 2). Because the molecule does not possess a chiral center, possible causes for the observed atropisomerism¹⁵ include hindered inversion of the sulfonamide nitrogen, restricted rotation about the N-S single bond, and conformationally constrained seven-membered diazepine ring system. On the basis of X-ray analysis of 8a and 8b, the bond distances from the sulfonamide N to the connected atoms are all typical for single-bond lengths, but the geometry around this N is almost planar (Figure 3). Both molecular modeling and X-ray analysis revealed that no obvious energy barrier existed for either the N-S bond rotation or the inversion of the nearly planar sulfonamide nitrogen. In contrast, the puckered diazepine ring was highly conformationally constrained, and a ring flip would place the sulfonyl group and the 7-Me substituent of the phenyl ring in the same plane, resulting in severe van der Waals interactions between these groups. Modeling suggested a barrier of \sim 33 kcal/mol for the ring interconversion. Chiral LC study of 8b afforded a first-order kinetics with a rate constant of 1.35×10^{-8} s⁻¹ at 25 °C and 1.03×10^{-7} s⁻¹ at 40 °C, corresponding to activation energy of 25.2 kcal/mol. Although the racemization is slow at <50 °C ($t_{1/2}$ = 296 days at 25 °C; 39 days at 40 °C), the rate of interconversion is significantly higher at >100 °C, and a complete racemization was observed after 4 h at 100 °C.¹⁶

The "SAR" of the atropisomerism is consistent with the planar chirality where the steric hindrance between the sulfonyl group and the neighboring substituent R_A on the Ph ring accounted for the interconversion barrier (Figure 4). The atropoisomers of the 7-H and 7-F analogues were not separable by HPLC at room temperature, neither was compound **9** with only an electron lone

Table 1. Substitution Effects on the Human and Mouse BRS-3 Activity



compd ^a	substitution	hBRS-3 ^{b} IC ₅₀ (nM)	hBRS-3 ^b EC ₅₀ (nM) (%Act) ^c	mBRS-3 ^b EC ₅₀ (nM) (% Act) ^c
1 7a	7-Me, 8-Me 9-Me, 10-Me	40 ± 21 80% inh ^d at 10 μ M	$325 \pm 117 (91 \pm 9\%)$ >10000 ^d	$248 \pm 42 (101 \pm 3\%)$ >10000 ^d
8a (S)	7-Me, 8-Me, 2-CF ₃	1.4 ± 0.9	$97 \pm 23 \ (96 \pm 10\%)$	$33 \pm 11 (98 \pm 7\%)$
8b (R)	7-Me, 8-Me, 2-CF ₃	169 ± 139	>10000	>10000
8c	7-Me, 8-Me, 2-Me	10 ± 4.5	$131 \pm 36 \ (92 \pm 13\%)$	$232 \pm 50 \ (107 \pm 10\%)$
8d (S)	7-Me, 8-Me, 2-CN	4.4 ± 1.0	$50 \pm 8 (100 \pm 4\%)$	$50 \pm 3 \ (103 \pm 1\%)$
8e (S)	7-Me, 8-Me, 2-Cl	1.2 ± 0.3	$38 \pm 9 (97 \pm 2\%)$	$98 \pm 25 \ (105 \pm 7\%)$
8f	7-Me, 8-Me, 3-Cl	430 ± 170	>10000 ^d	$> 10000^{d}$
8g	7-Me, 8-Me, 2-Cl, 3-F	8.9 ± 3.0	$277 \pm 80 (83 \pm 1\%)$	$371 \pm 13 \ (82 \pm 6\%)$
8h (S)	7-Me, 2-CF ₃	0.9 ± 0.1	$50 \pm 14 \ (101 \pm 10\%)$	$120 \pm 14 \ (106 \pm 1\%)$
8i (S)	7-Cl, 2-CF ₃	3.2 ± 0.5	$63 \pm 33 \ (96 \pm 6\%)$	$155 \pm 53 \ (100 \pm 13\%)$
8j	8-Me, 2-CF ₃	5.8 ± 0.1	$199 \pm 34 \ (83 \pm 11\%)$	$377 \pm 7 (82 \pm 4\%)$
8k	2-CF ₃	5.5 ± 0.5	$110 \pm 23 \ (96 \pm 10\%)$	$161 \pm 11 \ (98 \pm 7\%)$
81	7-F, 2-CF ₃ , 8-C(CH ₃) ₂ OH	5.0 ± 0.3	$77 \pm 10 \ (96 \pm 2\%)$	$134 \pm 14 \ (97 \pm 0\%)$
8m (S)	7-OH, 2-CF ₃	2.2 ± 0.4	$88 \pm 21 \ (103 \pm 1\%)$	$209 \pm 62 \ (115 \pm 14\%)$
9	see Figure 4	5.3 ± 0.2	$110 \pm 27 \ (104 \pm 0\%)$	$171 \pm 5 \ (112 \pm 12\%)$
10 (S)	see Figure 4	0.7 ± 0.1	$108 \pm 44 \ (101 \pm 8\%)$	$129 \pm 18 \ (107 \pm 4\%)$

^{*a*} Racemic mixture or achiral compound (7, 8j, 8k, etc.), except as noted. ^{*b*} Data expressed as mean \pm SD ($n \ge 2$ independent experiments). ^{*c*} % Act represents the maximum activation of tested compound relative to that of the dY peptide ([D-Tyr,⁷ β -Ala,¹⁷ Phe,¹⁹ and Nle²⁰]-bombesin(6–14)). ^{*d*} Assayed only once.

pair on R_A position. Not surprisingly, fused ring analogue **10** exhibited atropisomerism, and the two enantiomers were readily separated.

As expected, the two atropisomers, 8a and 8b, behaved very differently toward the BRS-3 receptors. The (S)-isomer 8a exhibited hBRS-3 IC₅₀ of 1.4 nM (125 I-dY-peptide was used as a radioligand in the binding assay) and EC_{50} of 97 nM, over 100-fold more potent than the (R)-isomer **8b** (Table 1).¹⁸ As compared to initial lead 1, CF3 at the 2-position (8a) improved hBRS-3 binding affinity by 30-fold and mouse functional activity by 8-fold. While the 2-position tolerated a variety of functionalities, such as methyl (8c), cyano (8d), and chloro (8e), the 3-position preferred no substituents (8f)vs 1; 8g vs 8e). For the left-hand side aromatic ring, deletion of 8-methyl (8h) slightly improved potency; however, removal of 7-methyl (8j) or both methyls (8k) decreased potency. Replacing the 7-methyl group with chloro (8i), fluoro (8l), or hydroxyl (8m) resulted in comparable potency. The fused ring analogue 10 was among the most potent compounds with hBRS-3 IC₅₀ of 0.7 nM.

Next, the SAR of the aryl sulfonyl group was explored.¹⁹ Analogues **11a**—**i** in Table 2 were synthesized as described in Scheme 1 using the appropriate sulfonyl chloride. para-Substituted phenyl was preferred, and a variety of substituents were tolerated. While isopropyl (**11a**) and trifluoromethoxy (**11d**) groups could serve as replacements for the *tert*-butyl (**8a**), the methylsulfonyl group (**11c**) caused loss of potency. In addition to phenyl analogues, heterocyclic analogues, such as thiazole (11g) and thiophene analogues (11h and 11i), also exhibited reasonable BRS-3 potency.

A number of more potent BRS-3 agonists were evaluated for their pharmacokinetic properties, and several displayed an improved PK profile (Table 3). As compared to the initial lead 1, these compounds demonstrated lower clearance and higher oral AUC and bioavailability in the rat. The pharmacokinetic properties of **8a** were also evaluated in the rhesus monkey where the bioavailability was 17% and the half-life was 1.8 h.

Compound 8a exhibited excellent potency against dog and rhesus BRS-3 receptors: dog $IC_{50} = 1.6$ nM, $EC_{50} = 36$ nM, 97% Act; rhesus $IC_{50} = 3.0$ nM, $EC_{50} = 70$ nM, 97% Act. This compound was screened against >100 receptors, ion channels, and enzymes; seven off-targets were identified with IC_{50} values in the range of $1-10 \ \mu$ M, but none was considered significant. Compound 8a was very selective over the two key ion channels, hERG $IC_{50} > 10 \ \mu$ M and DLZ $IC_{50} > 10 \ \mu$ M, but was a potent activator of human PXR ($EC_{50} = 678$ nM; 42% Act at $10 \ \mu$ M).²⁰

Compound **8a** was assessed for acute efficacy in wild-type (WT) and BRS-3 KO mice. In WT mice, the compound caused significant food intake reduction and body weight loss, in a dose-dependent manner at 3 and 30 mpk 18 h after oral dosing. In KO mice, no effect was observed at 3 and 30 mpk. After subchronic dosing (20 mpk, BID) in e-DIO mice, the compound caused 5.2% body weight loss on day 4 (Figure 5).²¹ The body weight loss was caused by reduction in food intake as well as increase in fasting metabolic rate.²²

Table 2. Human and Mouse BRS-3 Activity of 11



Cmpd ^a	Ar	hBRS- 3^{b} IC ₅₀ (nM)	hBRS- 3^{b} EC ₅₀ (nM) (%Act) ^c	mBRS-3bEC50 (nM)(%Act)c
11a (S)	2 C	1.6 ± 0.2	49 ± 16 (99 ± 5%)	84 ± 24 (110 ± 4%)
11b <i>(S)</i>	CF3	6.8 ± 1.8	154 ± 59 (78 ± 5%)	79 ± 15 (100 $\pm 2\%$)
11c	,z₂, SO₂Me	73 ± 21	$543 \pm 163 \\ (45 \pm 14\%)$	561 ± 55 (46 ± 8%)
11d <i>(S)</i>	JZ OCF3	3.3 ± 1.4	43 ±8 (101 ± 11%)	80 ± 14 (103 ± 2%)
11e	3 F	5.7 ± 3.4	69 ± 18 (94 ± 13%)	200 ± 20 (108 $\pm 2\%$)
11f (S)	2 CN	5.7 ± 2.4	60 ± 14 (80 ± 1%)	79 ± 2 (83 ± 2%)
11g	N	20 ± 5.7	216 ± 56 (96 ± 6%)	149 ± 13 (107 ± 2%)
11h	CF3	10 ± 1.9	158 ± 12 (84 ± 6%)	154 ± 21 (90 ± 1%)
11i	CF3	14 ± 2.1	123 ± 3 (93 ± 2%)	96 ± 14 (103 ± 2%)

^{*a*} Racemic mixture, except as noted. ^{*b*} Data expressed as mean \pm SD ($n \ge 2$ independent experiments). ^{*c*} % Act represents the maximum activation of tested compound relative to that of the dY peptide.

Table 3. Rat^a Pharmacokinetic Data^b of SelectedCompounds

compd	$Cl_p \ (mL/min/kg)$	$T_{1/2}$ (h)	po ${\rm AUC_N}~(\mu{\rm M~h~kg/mg})$	$F_{\rm oral}$ (%)
1	164	1.6	0.007	1.2
8a	29	1.6	0.20	15
8h	38	1.6	0.13	14
8i	46	1.4	0.26	34
8j	24	2.2	0.31	22
81	24	3.0	0.19	13
11a	16	3.5	0.24	10

 a Sprague—Dawley (SD) rat was used. b Plasma clearance (Cl_p) and half-life $(T_{1/2})$ calculated following 1 mg/kg iv dose. Normalized oral exposure (po AUC_N) and oral bioavailability ($F_{\rm oral}$) calculated following 4 mg/kg po dose.

In summary, we have discovered a novel class of benzodiazepine sulfonamide-based BRS-3 agonists along with the unusual planar chirality that originated from a conformationally restrained diazepine ring system. Starting from a high-throughput screening lead, SAR studies led to the synthesis of a series of BRS-3 agonists with improved potency and pharmacokinetic

Cumulative body weight change - Day 4





properties, of which compound 8a caused mechanism-based, dose-dependent food intake reduction and body weight loss after oral dosing in eDIO mice.

ASSOCIATED CONTENT

Supporting Information. Synthetic procedures and analytical data of selected BRS-3 agonists, conditions for all of the biological assays, and X-ray, chiral LC, and modeling methods. This material is available free of charge via the Internet at http:// pubs.acs.org.

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(15) Generally speaking, an atropisomer is a stereoisomer where the chiral element is not located on an atom but instead on a molecular plane or axis. The arbitrary definition of atropisomerism is that an atropisomer exists when the half life of interconversion is greater than 1000 s at room temperature. See Oki, M. Recent Advances in Atropisomerism. *Top. Stereochem.* **1983**, *14*, 1–81.

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(17) The assignment of absolute stereochemistry is based on the Cahn–Ingold–Prelog convention. The phenyldiamine plane is viewed as the chiral plane, and the sulfur is designated as the pilot atom; therefore, 8a has an (S)-configuration, and 8b has an (R)-configuration.

(18) There was a discrepancy between IC_{50} and EC_{50} because of the kinetic effects associated with the EC_{50} assay. For example, when **8a** was

assayed with a 10 min method, its EC_{50} value would be 15 nM. The longer the EC_{50} assay method is , the closer the IC_{50} and EC_{50} values are.

(19) The aliphatic sulfonamides and the corresponding *para-*^rbutylbenzoic acid amide (structures not shown) were not active.

(20) The human PXR issue will be addressed in a future communication.

(21) Bag-1 was a BRS-3 agonist from a different lead series; see refs 7 and 11 (compound 9 in ref 11). AM251 was a cannabinoid 1 receptor antagonist [see *Physiol. Behav.* 2004, 82 (5), 863–869] that was used here as a positive control.

(22) Compound **8a** was referred to as Bag-4 in this reference; see Metzger, J. M.; Gagen, K.; Raustad, K.; Yang, L.; White, A.; Wang, S.-P.; Craw, S.; Liu, P.; Lanza, T. J.; Lin, L. S.; Nargund, R. P.; Guan, X.-M.; Strack, A. M.; Reitman, M. L. Body temperature as a mouse pharmacodynamic response to bombesin receptor subtype-3 (BRS-3) agonists and other potential obesity treatments. *Am. J. Physiol. Endocrinol. Metab.* **2010**, *299*, 816–824.